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EAST Search History

S13	1328	potassium adj molybdate	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	SO.	NO	2007/11/13 08:57
2 4	1764	ammonium adj. heptamolybdate	US-PGPUB; USPAT; USOCR, EPO; JPO; DERWENT	OR	N O	2007/11/13 08:57
S15	13890	molybdenum adj oxide	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	S S	NO	2007/11/13 08:57
516	115	Si3 and Si4	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	ž	8	2007/11/13 08:58
517	64	S16 and S15	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	A S	NO	2007/11/13 11:25
818	3298	methylmercaptan; methanethio	US-PGPUB; USPAT; USOCK, EPO: JPO; DERWENT	Ř	NO	2007/11/13 11:49
819	1328	potassium adj molybdate	US-PGPUB; USPAT; USOCR; EPO; JPO; DERWENT	æ	NO	2007/11/13 11:25
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     FILE 'CASREACT' ENTERED AT 12:28:30 ON 13 NOV 2007
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L7
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L8
           7212 S AMMONIUM (A) MOLYBDATE
              1 S L3 AND L8
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L10
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              2 S L10 AND L8
L14
            273 S L10 AND L3
L15
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ANSWER 1 OF 1 REGISTRY COPYRIGHT 2007 ACS on STN 7

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766 L1

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FHITSTR' IS NOT A VALID FORMAT FOR FILE 'CASREACT' => d fhitstr 1-10

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All, AB. IND, RE, Single-step Reactions
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ABS, indented with text labels
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SIBIB ----- IBIB, no citations

FHITCELE --- FHIT, AND DIME GRIB
FCRD ----- First hit in Compact Reaction Display (CRD) format
FCRDREF ---- First hit in Compact Reaction Display (CRD) format with
CA reference information (SO, PY). (Default)
FPATH ----- PATH, plus Reaction Summary for the "short path"
FSPATH ----- REACTION MSD, Reaction Disgram, and Reaction
Summary for all hit reactions and fields containing Single-Step Reactions (Map, Diagram, and Summary for Same as ALL PI. SO PI. SO I and FCRD (random display, no answer number. SCA must be entered on the same line as DISPLAY, e.g., CRD ------ Compact Display of All Hit Reactions CRDREF ---- Compact Reaction Display and So, PY for Reference FHIT ----- Reaction Map, Diagram, and Summary for first all single-step reactions) BIB, IPC, and NCL hit reaction hit terms D SCAN.) STD -----

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ENTER DISPLAY FORMAT (FCRDREF):fcrdref

CASREACT COPYRIGHT 2007 ACS on ANSWER 1 OF 766 F3

REF: PCT Int. Appl., 2007085514, 02 Aug 2007
NOTE: autoclave used, heterogenous catalyst. 7-104 W03, 7-104Cs20, alumina support (Degussa)
CON: STAGE(1) room temperature -> 140 deg C; 2.5 hours, 140 deg C

L3 ANSWER 2 OF 766 CASREACT COPYRIGHT 2007 ACS on STN

RX(24) OF 73

(step 1)

75\$ 75\$ ters, 9(12), 2321-2324; 2007

Organic Letters, 9(12), 2321-2324; 2007
photochemical (uv), stereoselective (1somer mix.)
STARE(1) room temperature -> -30 deg C
STARE(2) 20 minutes, -30 deg C; 100 minutes, -30 deg C

L3 ANSWER 3 OF 766 CASREACT COPYRIGHT 2007 ACS on STN

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RX(7) OF 27

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MeSH, EtaN, THF, PhMe

70%

REF: Journal of Coordination Chemistry, 60(10), 1057-1067; 2007 CON: 4 days, room temperature

L3 ANSWER 4 OF 766 CASREACT COPYRIGHT 2007 ACS on STN

RX(3) OF 118

REF: Journal of Medicinal Chemistry, 50(9), 2067-2077; 2007 NOTE: methanethiol gas condensed into pressure bottle which is then sealed, molecular sieves used CON: STRGE(1) -40 deg C; 96 hours, room temperature

L3 ANSWER 5 OF 766 CASREACT COPYRIGHT 2007 ACS on STN

Page 6 11/13/07

Page 5 11/13/07

REF: Synthesis, (1), 55-60; 2007
NOTE: optimized on alkanethiol, optimization study
CON: STAGE(1) room temperature -> 0 deg C; 0 deg C; overnight,
0 deg C -> room temperature

L3 ANSWER 6 OF 766 CASREACT COPYRIGHT 2007 ACS ON STN

RX(1) OF S

REF: PCT Int. Appl., 2007032177, 22 Mar 2007 CON: STAGE(1) 0 deg C; 4 hours, 40 deg C, 0.25 MPa

L3 ANSWER 7 OF 766 CASREACT COPYRIGHT 2007 ACS on STN

RX(7) OF 291

Journal of the American Chemical Society, 129(4), 914-923; 2007 STAGE(1) -30 deg C STAGE(2) -30 deg C STAGE(2) -30 deg C STAGE(3) 0 deg C STAGE(3) 0 deg C.18 hours, room temperature 8EF::

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L3 ANSWER 8 OF 766 CASREACT COPYRIGHT 2007 ACS on STN

RX(2) OF 6

REF: PCT Int. Appl., 2007022900, 01 Mar 2007 NOTE: Reactant assumed CON: STAGE(1) 30 minutes, room temperature; 5 - 10 hours, 90 deg C

ANSWER 9 OF 766 CASREACT COPYRIGHT 2007 ACS on STN ជ

RX(2) OF 6

REF: PCT Int. Appl., 2007022901, 01 Mar 2007 NOTE: reactant assumed CON: STAGE(1) 30 minutes, room temperature; 5 - 10 hours, 90 deg C

L3 ANSWER 10 OF 766 CASREACT COPYRIGHT 2007 ACS on STN

RX(25) OF 115

$$clGt_2-Gt_2-Ct_2-Ct_2H \xrightarrow{1. \text{ MeSH}} \text{ Merg-Ct}_2-Gt_2-Ct_2H$$

REF: Bioorganic & Medicinal Chemistry Letters, 17(1), 73-77; 2007

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Page 8

(HYDROGEN OR HYDROGENS)

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5922 SULFIDES
15920 SULFIDE
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Page 9 11/13/07

10/595333 MERCAPTAN

(HYDROGEN OR HYDROGENS)

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CORPORATE SOURCE:

AUTHOR(S):

Novel Mo-based catalysts for methanethiol synthesis from high H2S-containing syngas Wang, Qi; Chen, Ai-Ping; Xie, Chun-Fang; Zheng, Quan-Xing; Fang, Wei-Ping; Yuan, You-Zhu; Zhang, Hong-Bin; Yang, Yi-Quan Department of Chemistry, Institute of Physical

L7 ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2007 ACS ON STN ACCESSION NUMBER: 2004:1130642 HCAPLUS DOCUMENT NUMBER: 142:431920

As series of supported Mo-based catalysts for methanethiol synthesis from high H2S-containing syngas were investigated by kinetics and KPS. ESR high H2S-containing syngas were investigated by kinetics and KPS. ESR characterization. The activity evaluating results show that upon the potassium-promotedMo-based catalysts, the methanethiol will become dominant product of the reaction, and the activity sequence of several Mo-based catalysts for the reaction, as follows: KZMOO4/COO/SIO2 > KZMOO4/SIO2 > WO3/KZOO3/SIO2 > KZMOO4/SIO2 > WO3/KZOO3/SIO2 > KZMOO4/SIO2 > WO3/KZOO3/SIO2 > KZMOO4/COO/SIO2 > WO3/KZOO3/SIO2 > KZMOO4/COO/SIO2 > MO3/KZOO3/SIO2 > MO3/KZOO3/ FILE 'HCAPLUS' ENTERED AT 12:31:07 ON 13 NOV.2007
0 S L3 AND HYDROGEN (W) HYDROGEN (A) SULFIDE (W) CARBON (A) OXIDE
137 S POTASSIUM (A) MOLYBDATE
1 S L6 AND L3 OXIDE Chemistry, Key Laboratory for Physical Chemistry of Solid Surface, Xiamen University, Xiamen, 361005, CARBON (A) Peop. Rep. China Huaxue Xuebao (2004), 62(23), 2297-2302 CODEN: HHFPA4; ISSN: 0567-7351 Kexue Chubanshe 'CASREACT' ENTERED AT 12:28:30 ON 13 NOV 2007
766 S L1
766 S L1 SSS FULL
76 S L3 AND HYDROGEN (W) HYDROGEN (A) SULFIDE (W) FILE 'REGISTRY' ENTERED AT 12:28:02 ON 13 NOV 2007 (FILE 'HOME' ENTERED AT 12:27:48 ON 13 NOV 2007) CASREACT 142:431920 (MOLYBDATE OR MOLYBDATES)
7212 AMMONIUM (A) MOLYBDATE (AMMONIUM OR AMMONIUMS) Journal Chinese => 6 ammonium (a) molybdate 405019 AMMONIUM 419 AMMONIUMS 405167 AMMONIUM 36812 MOLYBDATE 5695 MOLYBDATES 38616 MOLYBDATE DOCUMENT TYPE: LANGUAGE: OTHER SOURCE(S): AB A series of FILE PUBLI SHER => d his SOURCE:

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11/13/07 Page 11

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1337 S POTASSIUM (A) MOLYBDATE
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7212 S AMMONIUM (A) MOLYBDATE
1 S L3 AND L8

22728

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L9 ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 1998:288625 HCAPLUS DOCUMENT NUMBER: 129:54004

Effect on the reaction between methanol and hydrogen sulfide of Na or Mo doping on zirocnia and alumina Ziolek, M.; Kujawa, J.; Czyzniewska, J.; Nowak, I.; Aboulayt, A.; Saur, O.; Lavalley, J. C. AUTHOR(S):

Faculty of Chemistry, A. Mickiewicz University, Poznan, Pol. Paplied Catalysis, A. General (1998), 171(1), 109-115 CODEN: AcAGE4: ISSN: 0926-860X Elsevier Science B.V. CORPORATE SOURCE:

SOURCE:

PUBLISHER: DOCUMENT TYPE: LANGUAGE:

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The effect of sodium doping on alumina and zirconia activity in methanethiol formation from blumina and zirconia activity in methanethiol formation from hydrosulfurization of methanol by H2S was studied. It provoked an activity decrease and a selectivity increase in agreement with the catalysts basicity increase. By contrast, molybdenum doped zirconia samples presented a higher activity but a lower selectivity than pure zirconia, as expected from their higher acidity. Increase of the CH30H/H2S ratio in the reaction mixture improved both activity and CH3SH selectivity. It was found that zirconia loaded with small amts. of production
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6171 S METHYLMERCAPTAN OR METHANETHIOL OR METHANE (A) THIOL
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140 METHANE (A) THIOL
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2 L10 AND L8

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Yang, Yiquan, Wang, Qi; Lin, Renchun; Zhang, Hongbin; Yuan, Youzhu; Fang, Weiping; Zheng, Quanxing; Dai, Shenjun; Yan, Xingguo; Chen, Aiping; Barth, Jan-Olaf; Weckbecker, Christoph; Huthmacher, Klaus; Redlingshoefer, Hubert; Ackermann, Sabine Degussa, A.-G., Germany PCT Int. Appl., 31 pp. Process for the manufacture of methylmercaptan L12 ANSWER 1 OF 4 HCAPLUS COPYRIGHT 2007 ACS on STN 2005:395243 HCAPLUS 142:431977 PATENT ASSIGNEE(S): ACCESSION NUMBER DOCUMENT NUMBER: INVENTOR(S): SOURCE:

Patent English DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

20040915 CN 2003-10100495 20031010 20060051 CN 2003-10100496 20031010, 20060621 EP 2004-765676 20040312 20060621 EP 2004-765676 20040929 XK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, Y, TR, BG, CZ, EE, HU, PL, SK 20061125 CN 2004-80020637 20040929 20061205 BR 2004-15200 20040929 20070405 JP 2006-530037 20040929 20070405 US 2007-595333 20070126 CN 2003-10100495 A 20031010 CN 2004-100008377 N 200400312 CN 2004-100008377 N 200400329 ZW, AM, DE, DK, RO, SE, MR, NE, 20040929 표용감물 \$ £ £ £ DATE F 7 C 8 8 7 4 8 BY, KP, MX, BW, KG, MW, 8 F £ 7.8 WO 2004-EP10872 APPLICATION NO. AK, St. BB, DZ, IS, MG, S.T.A. AN NA CI, 20050506 AZ, DK, H. 8 H 7 K DATE ₩. GR., AU, DE, ć K AT, E.S. KIND 87.5.F.F. AA AA GU, 됐 당 , £, I AL GR, LS, TR, TR, W: AE, AG,
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SN, TD, R: AT, BE, IE, SI, PRIORITY APPLN. INFO. CN 1867545 BR 2004015200 JP 2007508256 US 2007213564 WO 2005040082 WO 2005040082 1528516 1559676 CN 1528515 CN 1528516 CN 1559676 EP 1670754 PATENT NO.

The present invention refers to a continuous process for the manufacture of Me mercaptan using Mo-O-K-based catalysts. It is further described that the total selectivity of methylmercaptan can be increased by at least 1% by lowering the total gas hourly space velocity. The invention further refers to a process for the preparation of a solid, preformed catalyst ΑB

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Page 13

system.

Novel Mo-based catalysts for methanethiol synthesis from high H2S-containing syngas Wang, Qi: Chen, Al-Ping; Xie, Chun-Fang; Zheng, Quan-Xing; Fang, Wei-Ping; Yuan, You-Zhu; Zhang, Hong-Bin; Yang, Yi-Quan Department of Chemistry, Institute of Physical Chemistry, Key Laboratory for Physical Chemistry, Key Laboratory for Physical Chemistry, Key Laboratory for Physical Chemistry of Solid Surface, Xiamen University, Xiamen, 361005, Peop. Rep. China
Huaxue Xuebao (2004), 62(23), 2297-2302
CODEN: HHHPA4; ISSN: 0567-7351 COPYRIGHT 2007 ACS on STN 2004:1130642 HCAPLUS CASREACT 142:431920 Journal Chinese HCAPLUS L12 ANSWER 2 OF 4 ACCESSION NUMBER: DOCUMENT TYPE: LANGUAGE: OTHER SOURCE(S): CORPORATE SOURCE: DOCUMENT NUMBER AUTHOR (S): PUBLI SHER: TITLE:

A series of supported Wo-based catalysts for methanethiol

A series of supported Wo-based catalysts for methanethiol

XPA. ESR characterization. The activity evaluating results show that upon
the potassium-promoted Mo-based catalysts, the methanethiol will

become dominant product of the reaction, and the activity sequence of
several Wo-based catalysts for the reaction is as follows: XXMOs4/COO/SiO2

> XZMOs4/CSIO2 > MOO3/XCOO/SiO2 > XZMOs4/CSIO2 > MOS2/XCOO/SiO2

> XZMOs4/CSIO2 > MOO3/XCOO/SiO2 > XZMOs4/CSIO2 > MOS2/XCOO/SiO2

> XZMOs4/CSIO2 > MOO3/XCOO/SiO2 > XZMOS4/CSIO2 > MOS2/XCOO/SiO2

> XZMOS4/CSIO3 > MOO3/XCOO/SiO2 > XZMOS4/CSIO2 > MOS2/XCOO/SiO2

> XZMOS4/CSIO3 > MOO3/XCOO/SiO2 > XZMOS4/CSIO2 > MOS2/XCOO/SiO2

> XZMOS4/CSIO3 > MOS3/XCOO/SiO2 > XZMOS4/CSIO2 > MOS2/XCOO/SiO2

> XZMOS4/CSIO3 > MOS3/XCOO/SiO2 > XZMOS4/CSIO2 > MOS2/XCOO/SiO2

> XZMOS4/CSIO3 > MOS3/XCOO/SiO2 > XZMOS4/CSIO2 > MOS3/XCOO/SiO2 > was proposed

Yang, Yi-Quan, Yang, Hua, Wang, Qi; Yu, La-Jia; Wang, Cheng; Dai, Shen-Jun; Yuan, You-Zhu
Department of Chemistry, Institute of Physical
Chemistry and State Key Laboratory for Physical
Chemistry of Solid Surfaces, Xiamen University, Study of the supported KZMoO4 catalyst for methanethiol synthesis by one step from high Xiamen, 361005, Peop. Rep. China Catalysis Letters (2001), 74(3-4), 221-225 CODEN: CALEER, ISSN: 1011-372X Kluwer Academic/Plenum Publishers Journal L12 ANSWER 3 OF 4 HCAPLUS COPYRIGHT 2007 ACS ON STN ACCESSION NUMBER: 2001:658931 HCAPLUS DOCUMENT NUMBER: 135:359357 H2S-containing syngas CORPORATE SOURCE: AUTHOR(S): PUBLI SHER: SOURCE:

11/13/07 Page 15

DOCUMENT TYPE:

10/595333 MERCAPTAN

LANGUAGE:

be controlled by the suitable choice of a catalyst. With activated Al203 as a standard support, incorporation of basic promotors such as oxides and salts and alkaline earth oxides increased the RSH formation, whereas acid promotors such as HBPO4 increased the R25 formation. Comparison of activated Al203, various compas. of SiO2-Al203 and SiO2-MGO as supports showed an inverse relation between RSH production and KOH acidity. The selectivity of the catalyst and yield of RSH depended on the concentration of promotors and catalyst temperature MO03/K2CO3/SiO2 and K2MOO4/SiO2 catalysts were characterized and evaluated in MeSH preparation from high H2S-containing syngas. The two catalysts have similar activity in the reaction.

19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RENGE COUNT: The formation of RSH or R2S in the high-temperature reaction of ROH with H2S RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT 59:3218
59:825b-c
The role of the catalyst in the reaction of alcohols and hydrogen sulfide
Polkins, Hills 0.; Miller, Blmer L.
Pure Oil Co., Crystal Lake, IL
Proc. Am. Petrol. Inst. (1962), Sect. III 42, 188-96 L12 ANSWER 4 OF 4 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 1963:432418 HCAPLUS Unavailable Journal DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: AUTHOR(S): CORPORATE SOURCE: REFERENCE COUNT DOCUMENT TYPE: LANGUAGE: AB The. f can SOURCE TITLE: AB

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FILE 'HCAPLUS' ENTERED AT 12:31:07 ON 13 NOV 2007

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137 S POTASSIUM (A) MOLYBDATE

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7212 S AMMONIUM (A) MOLYBDATE

1 S L3 AND L8

6171 S METHYLMERCAPTAN OR METHANETHIOL OR METHANE (A) THIOL

7216 S L10 AND L6

2 S L10 AND L6

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L13 ANSWER 1 OF 2 HCAPLUS COPYRIGHT 2007 ACS on STN ACCESSION NUMBER: 1999:288625 HCAPLUS DOCUMENT NUMBER: 129:54004

Page 16 11/13/07

σ T P P P I L14 L5 L6 L9 L10 L11 L12 AU ٨ ፤ SS 222 AB The effect of sodium doping on alumina and zirconia activity in mechanethiol formation from hydrosulfurization of methanol by H2S mechanethiol formation from hydrosulfurization of methanol by H2S was studied. It provoked an activity decrease and a selectivity increase in agreement with the catalysts basicity increase. By contrast, melybdenum doped zirconia samples presented a higher activity but a lower selectivity than pure zirconia, as expected from their higher acidity. Increase of the GH3OH/H2S ratio in the reaction mixture improved both activity and CH3SH selectivity. It was found that zirconia loaded with small ante, of molybdenum working under H2S excess shows good performances towards GH3SH production.

THERE ARE ARE 26 CITED REFERENCES AVAILABLE FOR THIS THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT Effect on the reaction between methanol and hydrogen sulfide of Na or Mo doping on zirconia and alumina Ziolek, M.; Kujawa, J.; Cyzniewska, J.; Nowak, I.; Aboulayt, A.; Suur, O.; Lavalley, J. C. Faculty of Chemistry, A. Mickiewicz University, Poznan, Pol.
Applied Catalysis, A. General (1998), 171(1), 109-115 CODEN: RAGES; ISSN: 0926-860X 19750201 19760128 Alkylaufonic acids
Alkylaufonic acids
Schreyer, Gerd; Geiger, Friedhelm; Hensel, Joerg
Deutsche Gold- und Silber-Scheideanstaltvorm.
Roessler, Red. Rep. Ger.
Ger. Offen., 7 pp.
CODEN: GWXXBX DE 1975-2504201 FR 1976-2324 APPLICATION NO. J976:576836 HCAPLUS CASREACT 129:54004 19760819 DATE English Patent German KIND A1 B1 A1 A HCAPLUS LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: L13 ANSWER 2 OF 2 ACCESSION NUMBER: INVENTOR(S): PATENT ASSIGNEE(S): DOCUMENT TYPE: LANGUAGE: OTHER SOURCE(S): AB The effect of CORPORATE SOURCE: DOCUMENT NUMBER: DE 2504201 FR 2299319 FR 2299319 BE 838148 US 4052445 PATENT NO. DOCUMENT TYPE: AUTHOR(S): PUBLI SHER SOURCE:

H2O2 at 10.90° gave 85.5% MeSO3H. Without I there was no reaction.

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7216 S L10 AND L6 OR L8

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Effect on the reaction between methanol and hydrogen sulfide of Na or Mo doping on zirconia and alumina Ziolek, M.; Kujawa, J.; Czyzniewska, J.; Nowak, I.; Aboulayt, A.; Saur,
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142:431920 Novel Mo-based catalysts for methanethiol synthesis from high COPYRIGHT 2007 ACS on STN HCAPLUS ANSWER 1 OF 1 HCAPLUS 2004:1130642 HCAPLUS 11 Se E15

H2S-containing syngas Wang, Qi; Chen, Ai-Ping; Xie, Chun-Fang; Zheng, Quan-Xing; Fang, Wei-Ping;

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Yuan, You-Zhu; Zhang, Hong-Bin; Yang, Yi-Quan Department of Chemistry, Isolad Structure of Physical Chemistry, Key Laboratory for Physical Chemistry of Solid Surface, Xiamen University, Xiamen, 36105, Peop. Rep. China Huaxue Xuebao (2004), 62(23), 2297-2302
CODEN: HHHPA; ISSN: 0567-7351
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222

766 S L1 766 S L1 SSS FULL 0 S L3 AND HYDROGEN (W) HYDROGEN (A) SULFIDE (W) CARBON (A) OXIDE

FILE 'HCAPLUS' ENTERED AT 12:31:07 ON 13 NOV 2007

0 S.13 AND HYDROGEN (W) HYDROGEN (A) SULFIDE (W) CARBON (A) OXIDE

137 S POTASSIUM (A) MOLYBDATE

1 S. L6 AND L3

7212 S AMONIUM (A) MOLYBDATE

1 S.13 AND L8

6171 S METHYLMERCAPTAN OR METHANETHIOL OR METHANE (A) THIOL

7216 S. L10 AND L6

2 S. L10 AND L6

2 S. L10 AND L6

1 S. L14 AND L6

1 S. L14 AND L6

1 S. L14 AND L6 L5 L6 L7 L10 L11 L12 L13 L13

FILE 'STNGUIDE' ENTERED AT 12:43:05 ON 13 NOV 2007

=> d his

(FILE 'HOME' ENTERED AT 12:27:48 ON 13 NOV 2007)

FILE 'REGISTRY' ENTERED AT 12:28:02 ON 13 NOV 2007 1 S 74-93-1/RN ፫

FILE 'CASREACT' ENTERED AT 12:28:30 ON 13 NOV 2007
766 S L1
766 S L1 SSS FULL
0 S L3 AND HYDROGEN (W) HYDROGEN (A) SULFIDE (W) CARBON (A) OXIDE L3 L2

FILE 'HCAPLUS' ENTERED AT 12:31:07 ON 13 NOV 2007

0 \$ 1.3 AND HYDROGEN (M) HYDROGEN (A) SULFIDE (W) CARBON (A) OXIDE
133 \$ \$ POTASSIUM (A) MOLYBDATE
1 \$ 1.6 AND 1.3

7.215 \$ AMYONIUM (A) MOLYBDATE
6171 \$ AMYONIUM (A) MOLYBDATE
6171 \$ METHYLMERCAPTAN OR METHANETHIOL OR METHANE (A) THIOL
7216 \$ 1.10 AND 1.6

2 \$ 1.10 AND 1.6

2 \$ 1.10 AND 1.6

1 \$ 1.14 AND 1.6

1 \$ 1.14 AND 1.6 LS L6 L7 L10 L110 L111 L113 L114 L115

FILE 'STNGUIDE' ENTERED AT 12:43:05 ON 13 NOV 2007

FILE 'HCAPLUS' ENTERED AT 12:46:56 ON 13 NOV 2007

1 S YANG, Y?/AU AND WANG, Q?/AU AND LIN, R?/AU AND ZHANG, H?/AU AND Y

0 S BARTH, J?/AU AND WECKERBECKER, C?/AUAND HUTHWACHER, K?/AUAND R L17 L18

=> \$ 17 and 118 L19 0 L7 AND L18